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4 Problems and Solutions in Atomic, Nuclear and Particle Physics Solution:

The Bohr radius of the hydrogen atom and the Compton wavelength of electron are given by $a = \frac{2\pi\epsilon_0\hbar^2}{me^2}$ and $\lambda_c = \frac{h}{mc}$ respectively. Hence $\lambda_c = \frac{2\pi\epsilon_0\hbar^2}{mc^2} = \frac{2\pi}{137} \frac{\hbar}{mc} = \frac{2}{137} \lambda_c$, where $\lambda_c = \frac{h}{mc}$ is the fine-

structure constant. Hence the answer is (a). 1004
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Answer: $\rho = 737 \text{ kg/m}^3$; 1 gal = 3.788 L = $3.788 \times 10^{-3} \text{ m}^3$. The mass of a ^{235}U nucleus is $235m_p$, then. (a) for 1 gallon, $m = \rho V = (737 \text{ kg/m}^3)(3.788 \times 10^{-3} \text{ m}^3) = 2.79 \text{ kg} = 2.79 \times 10^{-3} \text{ g}$. So, $(1.3 \times 10^8) / (2.79 \times 10^{-3} \text{ g}) = 4.7 \times 10^5 / \text{g}$. (b) 1 g contains $(1.00 \times 10^{-3} \text{ kg}) / (235m_p) = 2.55 \times 10^{21}$ nuclei.

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