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8 Solutions Manual • Fluid Mechanics, Fifth Edition. In order for  $f_{cn}(H)$  to provide dimensions of  $\{L^3/2\}$ , the function must be a  $3/2$  power. Thus the final desired homogeneous relation for dam flow is:  $Q = CBg^{1/2}H^{3/2}$ , where  $C$  is a dimensionless constant. Ans.

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Solutions Manual Fluid Mechanics, Seventh Edition From Table A.3, methanol has  $\rho = 791 \text{ kg/m}^3$  and a large vapor pressure of 13,400 Pa. Then the manometer rise  $h$  is given by  $p_L = \rho g h$ .  $p_{\text{atm}} - p_{\text{vapor}} = \rho g h$ .  $h = \frac{p_{\text{atm}} - p_{\text{vapor}}}{\rho g} = \frac{13400 - 101325}{791 \cdot 9.81} = -11.8 \text{ m}$ . Solve for

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Note: Unit cancellations are not shown in this solution. a.) Situation: Pressure varies with elevation.  $z = 6.8 \text{ ft}$ . Find: Pressure change (kPa). Properties:  $\rho = 90 \text{ lbfm} = \text{ft}^3$ . Solution: Convert density to units of  $\text{kg/m}^3$ :  $\rho = 90 \text{ lbfm} = \text{ft}^3 \cdot 35.315 \text{ ft}^3 \text{ m}^{-3} = 1.0 \text{ kg} \cdot 2.2046 \text{ lbf} = 1442 \text{ kg} \cdot 9.81 \text{ m} \cdot \text{s}^2 = 6.8 \text{ ft} \cdot 1.0 \text{ m} \cdot 3 \dots$

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Solutions Manual for Fluid Mechanics: Fundamentals and Applications Third Edition Yunus A. Çengel & John M. Cimbala McGraw-Hill, 2013 CHAPTER 1 INTRODUCTION AND BASIC CONCEPTS PROPRIETARY AND CONFIDENTIAL This Manual is the proprietary property of The McGraw-Hill Companies, Inc.

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SOLUTION If we consider the dam as having a width of  $b = 1 \text{ m}$ , the intensity of the distributed load at the base of the dam is  $w_b = \rho g h b = 11000 \text{ kg} \cdot \text{m}^3 \cdot 219.81 \text{ m} \cdot \text{s}^2 = 219 \text{ m}^2 \cdot \text{s}^2 = 88.291103 \text{ kN} \cdot \text{m}^{-2}$

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Fluid Mechanics, 6th Ed. Kundu, Cohen, and Dowling Exercise 1.3.3 The Maxwell probability distribution,  $f(v) = f(v_x, v_y, v_z)$ , of molecular velocities in a gas flow at a point in space with average velocity  $u$  is given by (1.1). a) Verify that  $u$  is the average molecular velocity, and determine the standard deviations ( $\sigma_x, \sigma_y, \sigma_z$ ).

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