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Tension. Vaporization.
the attraction of the
surface of a liquid to
the surface of a.... a
substance that can
flow and therefore take
the shape of its.... a
force that tends to pull
adjacent parts of a
liquid's surfac....

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in this set (23) soluble.

Capable of being

dissolved. solution. A

homogeneous mixture

of 2 or more

substances in a single

phase. solvent. The

dissolving medium in a

solution.

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capable of dissolving in
a particular solvent.

solution. a

homogeneous mixture

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of two or more substances uniformly dispersed throughout a single phase.

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Chapter Test Name

_____ Class _____ Date

_____ Chapter Test A,
continued _____ 11.

Which of the following
is soluble in water? a.

potassium nitrate b.

silver c. benzene d.

carbon tetrachloride

_____ 12. In a solution

at equilibrium, a. no

dissolution occurs. b.

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Modern Chemistry 91
Chapter Test Name
Class Date Chapter
Test B, continued 27.
Distinguish between
ionic crystals and
metallic crystals. PART
V Write the answers to
the following questions
on the line to the left,
and show your work in
the space provided.

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Modern Chemistry Chapter 12 Review Answer Key

Modern Chemistry 1
Solutions CHAPTER 12
REVIEW Solutions
Teacher Notes and
Answers Chapter 12
SECTION 1 SHORT
ANSWER 1. c 2. a 3. b
2. a. alcohol b. water c.
the gels 3. The mixture
is a colloid. The
properties are
consistent with those
reported in Table 3 on

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page 404 of the text.

The particle size is small, but not too small, and the mixture

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Solutions and
Colligative Properties
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CHAPTER 12
HOMEWORK 12-5 (pp.
379-381) ...

STANDARDIZED TEST
PREP Circle the letter
of the best answer. 1.
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Particle Mass number
19. Proton 20. Neutron
21. Electron 10

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Relative charge

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ANSWER Write the

answers to the

following questions in

the space provided. 22.

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