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The salt bridge usually consists of a strong electrolyte which is further made up of ions. For example, AgNO₃, KCl, etc. Salt bridges are generally used in a galvanic cell such as a voltaic cell or Daniel cell. Salt Bridge Function. The main function of a salt bridge is to help maintain the electrical neutrality within the internal circuit.

Salt Bridge - Definition, Function, Types, Preparation ... - BYJUS

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Gauss law states that the total amount of electric flux passing through any closed surface is directly proportional to the enclosed electric charge. The applications of Gauss Law are mainly to find the electric field due to infinite symmetries such as:. Uniformly charged Straight wire; Uniformly charged Infinite plate sheet; Uniformly charged thin spherical shell

Applications of Gauss's Law: Overview, Formula and Derivations

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